

Directing Group Assisted Nucleophilic Substitution of Propargylic Alcohols via o-Quinone Methide Intermediates: Brønsted Acid Catalyzed, Highly Enantio- and Diastereoselective Synthesis of 7-Alkynyl-12a-acetamido-Substituted Benzoxanthenes

Satyajit Saha and Christoph Schneider*

Institute of Organic Chemistry, University of Leipzig, Johannisallee 29, D-04103 Leipzig, Germany

Supporting Information

ABSTRACT: BINOL-based, chiral phosphoric acids catalyze the substitution of 1-(o-hydroxyphenyl)propargylic alcohols with enamides to furnish 7-alkynyl-12a-acetamido-substituted benzo[c]xanthenes and related heterocycles in a one-pot operation with excellent diastereo- and enantioselectivity. Ambient reaction temperature, operationally simple reaction conditions, low catalyst loading, high yields, and excellent stereocontrol are attractive features of this process and make it a highly practical and versatile transformation.

C tudies on the direct nucleophilic displacement of the hydroxyl group in propargylic alcohols are currently among the prime objectives in organic synthesis. The highly versatile nature of the alkyne moiety makes such reactions highly attractive in organic synthesis. Fully or semisaturated products are easily obtained by catalytic hydrogenation, and various other functional groups are readily accessible by a range of known transformations of the alkynyl group.²

Accordingly, transition-metal-catalyzed propargylic substitution reactions have attracted a great deal of attention recently.³ Gold-, ruthenium-, and copper-catalyzed processes have been disclosed by the groups of Campagne, ⁴ Toste, ⁵ Sheppard, ⁶ Nishibayashi, ⁷ Alexakis, and others, respectively. ⁸ In addition, Nishibayashi et al. have disclosed powerful ruthenium- and copper-catalyzed enantioselective alkylation and amination reactions of propargylic alcohols and derivatives thereof.5 They have also developed cooperative ruthenium-prolinol ether- and ruthenium-copper-catalyzed asymmetric reactions of propargylic alcohols and esters. Recently, Hu¹¹ and van Maarseeven 12 and co-workers have reported various coppercatalyzed, enantioselective substitution reactions of propargylic acetates as well. Nevertheless, there is a great demand for novel, enantioselective propargylic substitution reactions that are applicable to a broad substrate range.

Recently, our group has developed a series of highly efficient Brønsted acid catalyzed, enantioselective substitution reactions of o-hydroxybenzhydryl alcohols employing 1,3-dicarbonyl compounds, enamides, indoles, and naphthols as nucleophiles. 13 Pursuing this strategy, we have successfully established one-pot and high-yielding syntheses of various chromene- and xanthene-based oxygen heterocycles as well as triarylmethanes with excellent stereocontrol. These transformations proceeded through in situ generated o-quinone methides (o-QM) activated through the hydrogen-bonded chiral phosphoric acid catalyst. 14 o-QMs are highly reactive 1-oxabutadienes which have frequently been employed in the synthesis of natural products and bioactive molecules. 15 However, the transient nature of o-QM principally poses challenges to synthetic applications, and it was only recently that a range of catalytic, enantioselective processes have been successfully developed for o-QM chemistry including palladium, cinchona alkaloid, BINOL, Brønsted acid, and NHC catalyzed reactions. 16

We report herein the Brønsted acid catalyzed, highly enantioselective substitution of 1-(o-hydroxyphenyl)propargylic alcohols 1 with enamides 2 which upon subsequent N,Oacetalization generate highly functionalized 7-alkynyl-12aacetamido-substituted benzo[c]xanthenes 4 and related heterocycles carrying three contiguous chiral centers. The products were obtained in good yield and excellent diastereo- and enantioselectivity in a one-pot operation (Scheme 1b). The ohydroxy group within 1 helps to form the highly reactive o-QM, which readily adds the enamide nucleophile in a conjugate

Received: December 20, 2014 Published: January 22, 2015

Organic Letters Letter

Scheme 1. Brønsted Acid Catalyzed Reaction of Propargylic Alcohols (A) without and (B) with Directing Group via o-QM

addition event. Accordingly, substrates lacking this o-hydroxy group or carrying a protected o-hydroxy group (e.g., OMe) do not react under these conditions (Scheme 1a).

We initiated our studies by investigating the reaction of 1-(o-hydroxyphenyl)-3-phenylpropargylic alcohol (1a) (1 equiv) with enamide 2a (1 equiv) in CH_2Cl_2 in the presence of various chiral phosphoric acids 3a-e (5 mol %) (Table 1). 7-Alkynyl-

Table 1. Optimization Studies a-d

entry	cat.	solvent	time (h)	% yield ^b	dr ^c	er^d
1	3a	CH_2Cl_2	8	64	>95:5	96:4
2	3b	CH_2Cl_2	16	50 ^c	>95:5	99:1
3	3c	CH_2Cl_2	8	72	>95:5	99:1
4	3d	CH_2Cl_2	8	70	>95:5	98:2
5	3e	CH_2Cl_2	16	57	96:4	90:10

^aAll reactions were carried out with 0.23 mmol (1 equiv) of 1, 0.23 mmol (1.0 equiv) of enamide 2a, and 5 mol % of catalyst 3a-e in 2 mL of CH_2Cl_2 at rt for 8-16 h. ^bIsolated yield of the product. ^cdr was determined through 1H NMR analysis of the crude reaction mixture. ^der was determined through chiral HPLC analysis (see the Supporting Information).

12a-acetamido-substituted benzo[c]xanthene 4a was obtained as largely a single diastereomer in good to moderate yields in almost all cases studied within 8-16 h at room temperature. The results further revealed that ortho-disubstitution within the 3,3'-aryl substituents in the BINOL backbone of the Brønsted acid catalyst gave superior enantioselectivities. For example, with phosphoric acids 3a and 3b carrying mesityl and 2,4,6triisopropyl phenyl groups as 3,3'-substituents, the product was obtained with excellent selectivity but only moderate yield (entries 1 and 2). Phosphoric acid 3d gave rise to an improved yield while maintaining the excellent enantioselectivity observed before (entry 4). Phosphoric acid catalyst 3c with pentamethylated 3,3'-phenyl groups was found to be the optimal catalyst for this reaction, which gave the highest enantioselectivity of 99:1 er in combination with excellent diastereoselectivity of >95:5 and good product yield of 72%

(entry 3). Neither yield nor stereoselectivity of the reaction were affected through the use of molecular sieves.

To assess the applicability of this process, various 1-(o-hydroxyphenyl)propargylic alcohols 1a—l were subsequently reacted with enamide 2a under the above-optimized reaction conditions, and the results are summarized in Table 2. In all

Table 2. Substrate Scope a-d

"All reactions were carried out with 0.42 mmol (1 equiv) of 1, 0.42 mmol (1.0 equiv) of enamide 2a, and 5 mol % of catalyst 3c in 4 mL of CH₂Cl₂ at rt for 8–36 h. ^bIsolated yield of the purified product. ^cdr was determined by ¹H NMR analysis of the crude reaction mixture. ^der was determined through chiral HPLC analysis (see the Supporting Information).

cases studied, the reaction proceeded smoothly and was typically completed within 8–36 h at rt. Inspection of Table 2 reveals that the products were obtained typically in good yield and excellent diastereo- and enantioselectivity, irrespective of the substitution pattern within the alkynyl or the aryl group. A broad range of internal alkynes with various substituents like aromatic, carbocyclic, ethers, aliphatic, long-chain aliphatic, and TMS moieties were readily tolerated, delivering the products with excellent results as well. Even a terminal alkyne withstood the reaction conditions and furnished the desired product 41 in moderate yield but with excellent diastereo- and enantioselectivity. A crystal structure of 4e (Table 2) and 7 (Table 3) revealed the absolute configuration which was assigned to all other products as well (see the SI for details). 17

In order to further broaden the scope of this reaction with respect to the enamide component, we employed some more carbocyclic as well as heterocyclic enamides in this process, all Organic Letters Letter

Table 3. Substrate $Scope^{a-d}$

^aAll reactions were carried out with 0.42 mmol (1.0 equiv) of 1, 0.42 mmol (1.0 equiv) of enamides 2b–f, and 5 mol % of catalyst 3c in 4 mL of CH₂Cl₂ at rt for 8–36 h. ^bIsolated yield of the purified product. ^cdr was determined by ¹H NMR analysis of the crude reaction mixture. ^der was determined through chiral HPL analysis (see the Supporting Information).

of which delivered the desired products in good yield and excellent selectivity (Table 3).

Quite interestingly, using chromene- and thiochromene-based enamides **2c** and **2d** as nucleophiles furnished the corresponding alkynyl-substituted chromeno[4,3-*b*]chromenes **6a**—c and thiochromeno[4,3-*b*]chromene 7, respectively, in good yield, exceptional diastereo- and enantiocontrol. Enamide **2e**, lacking the annulated aromatic ring, also conferred the addition product **8** in good yield, diastereoselectivity, and **88**:12 er. With regioisomeric enamide **2f** as substrate, alkynylated benzo[*a*]xanthene **9** was successfully obtained, albeit with decreased selectivity. Acyclic enamides, however, currently give rise to only low enantioselectivity. ¹⁸

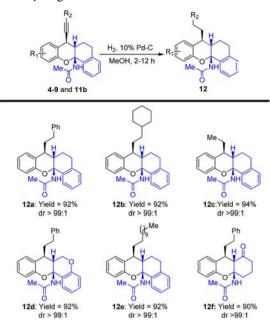
The excellent results obtained with enamides 2a-f encouraged us to further investigate *N*-acetylated enaminone 11 as a nucleophile. Under the optimized conditions, 11 underwent nucleophile substitution with propargylic alcohols 1a and 1h followed by *N*,*O*-acetalization to yield the desired products 11a and 11b, respectively, in good yield and moderate selectivity (Scheme 2).

To further reveal the synthetic potential of this new process, some of the products were subsequently subjected to catalytic hydrogenation using 10% Pd/C to furnish 7-alkyl-12a-acetamido-substituted tetrahydrobenzo[c]xanthenes 12a-f with facility (Table 4). This strategy offers an excellent method of accessing these products which otherwise would be inaccessible.

In summary, we have developed a highly enantioselective, Brønsted acid catalyzed propargylic alkylation of 1-(o-hydroxyphenyl)propargylic alcohols with enamides, a process

Scheme 2. Enaminone 10 as Nucleophile

Table 4. Hydrogenation Reactions a-c



"All reactions were carried out with 0.42 mmol (1 equiv) of 4-9 and 12b, 10 mg of Pd/C catalyst in 6 mL of CH₃OH at rt under positive pressure of H₂ for 2–12 h. ^bIsolated yield of the purified product. ^cdr was determined through ¹H NMR analysis of the product mixture.

which upon subsequent N,O-acetalization furnishes a broad range of 7-alkynyl-12a-acetamido-substituted benzo[c]-xanthenes and related heterocycles efficiently. The products can be easily converted into their highly valuable saturated analogues by catalytic hydrogenation. This study further emphasizes the efficacy of phosphoric acid catalyzed, enantioselective reactions of o-QMs and extends the scope of synthetic transformations involving enamide nucleophiles.

ASSOCIATED CONTENT

S Supporting Information

Detailed experimental procedures, spectral data for all new compounds, and crystallographic data and CIF information for 4e and 7. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: schneider@chemie.uni-leipzig.de.

Notes

The authors declare no competing financial interest.

Organic Letters Letter

■ ACKNOWLEDGMENTS

We are grateful to the Deutsche Forschungsgemeinschaft for the generous financial support of this study. We thank Dr. P. Lönnecke (University of Leipzig) for obtaining the crystalstructure analysis. We gratefully acknowledge the donation of chemicals from Evonik and BASF.

REFERENCES

- (1) Reviews: (a) Ljungdahl, N.; Kann, N. Angew. Chem., Int. Ed. 2009, 48, 642. (b) Detz, R. J.; Hiemstra, H.; van Maarseveen, J. H. Eur. J. Org. Chem. 2009, 6263. (c) Miyake, Y.; Uemura, S.; Nishibayashi, Y. Chem. Catal. Chem. 2009, 1, 342. (d) Ding, C.-H.; Hou, X.-L. Chem. Rev. 2011, 111, 1914. (e) Nishibayashi, Y. Synthesis 2012, 489. (f) Bauer, E. B. Synthesis 2012, 44, 1131.
- (2) Reviews: (a) Pohlki, F.; Doye, S. Chem. Soc. Rev. 2003, 32, 104. (b) Nunez, E. J.; Echavarren, A. M. Chem. Commun. 2007, 333. (c) Villar, H.; Fringsa, M.; Bolm, C. Chem. Soc. Rev. 2007, 36, 55. (d) Zani, L.; Bolm, C. Chem. Commun. 2006, 4263. (e) Chinchilla, R.; Najera, C. Chem. Soc. Rev. 2011, 40, 5084. (f) Song, G.; Wang, F.; Li, X. Chem. Soc. Rev. 2012, 41, 3651. (g) Fürstner, A. Angew. Chem., Int. Ed. 2013, 52, 2794. (h) Sokolova, N. V.; Nenajdenko, V. G. RSC Adv. 2013, 3, 16212. (i) Yamamoto, Y. Chem. Soc. Rev. 2014, 43, 1575. (j) Chinchilla, R.; Najera, C. Chem. Rev. 2014, 114, 1783.
- (3) Reviews: (a) Bruce, M. I. Chem. Rev. 1998, 98, 2797. (b) Bruneau, C.; Dixneuf, P. H. Angew. Chem., Int. Ed. 2006, 45, 2176. (c) Cadierno, V.; Gimeno, J. Chem. Rev. 2009, 109, 3512. (d) Yan, W.; Wang, Q.; Chen, Y.; Petersen, J. L.; Shi, X. Org. Lett. 2010, 12, 3308. (e) Sperger, C. A.; Tungen, J. E.; Fiksdahl, A. Eur. J. Org. Chem. 2011, 3719. (f) Garayalde, D.; Kruger, K.; Nevado, C. Angew. Chem., Int. Ed. 2011, 50, 911. (g) Ding, C.-H.; Hou, X.-L Chem. Rev. 2011, 111, 1914.
- (4) (a) Georgy, M.; Boucard, V.; Campagne, J.-M J. Am. Chem. Soc. 2005, 127, 14180. Review: (b) Georgy, M.; Boucard, V.; Debleds, O.; Dal Zotto, C.; Campagne, J.-M. Tetrahedron 2009, 65, 1758. (c) Debleds, O.; Gayon, E.; Vrancken, E.; Campagne, J.-M. Beilstein J. Org. Chem. 2011, 7, 866.
- (5) Shapiro, N. D.; Shi, Y.; Toste, F. D. J. Am. Chem. Soc. 2009, 131, 11654.
- (6) Pennell, M. N.; Turner, P. G.; Sheppard, T. D. Chem.—Eur. J. 2012, 18, 4748.
- (7) (a) Nishibayashi, Y.; Yoshikawa, M.; Inada, Y.; Hidai, M.; Uemura, S.; Sakae. J. Am. Chem. Soc. 2002, 124, 11846. (b) Nishibayashi, Y.; Yoshikawa, M.; Inada, Y.; Hidai, M.; Uemura, S. J. Org. Chem. 2004, 69, 3408. (c) Nishibayashi, Y.; Milton, M. D.; Inada, Y.; Yoshikawa, M.; Wakiji, I.; Hidai, M.; Uemura, S. Chem.—Eur. J. 2005, 11, 1433.
- (8) (a) Fang, P.; Hou, X.-L. Org. Lett. 2009, 11, 4612. (b) Li, H.; Grassi, D.; Gune, L.; Brgi, T.; Alexakis, A. Chem.—Eur. J. 2014, 20, 16694. (c) Zhao, L.; Huang, G.; Guo, B.; Xu, L.; Chen, J.; Cao, W.; Zhao, G.; Wu, X. Org. Lett. 2014, 16, 5584.
- (9) (a) Nishibayashi, Y.; Imajima, H.; Onodera, G.; Uemura, S. Organometallics 2005, 24, 4106. (b) Inada, Y.; Nishibayashi, Y.; Uemura, S. Angew. Chem., Int. Ed. 2005, 44, 7715. (c) Matsuzawa, H.; Miyake, Y.; Nishibayashi, Y. Angew. Chem., Int. Ed. 2007, 46, 6488. (d) Matsuzawa, H.; Kanao, K.; Miyake, Y.; Nishibayashi, Y. Org. Lett. 2007, 9, 5561. (e) Fukamizu, K.; Miyake, Y.; Nishibayashi, Y. J. Am. Chem. Soc. 2008, 130, 10498. (f) Hattori, G.; Matsuzawa, H.; Miyake, Y.; Nishibayashi, Y. Angew. Chem., Int. Ed. 2008, 47, 3781. (g) Kanao, K.; Miyake, Y.; Nishibayashi, Y. Organometallics 2009, 28, 2920. (h) Hattori, G.; Sakata, K.; Matsuzawa, H.; Tanabe, Y.; Miyake, Y.; Nishibayashi, Y. J. Am. Chem. Soc. 2010, 132, 10592. (i) Kanao, K.; Miyake, Y.; Nishibayashi, Y. Organometallics 2010, 29, 2126. (j) Hattori, G.; Miyake, Y.; Nishibayashi, Y. Chem. Catal. Chem. 2010, 2, 155. (k) Shibata, M.; Nakajima, K.; Nishibayashi, Y. Chem. Commun. 2014, 50, 7874.
- (10) (a) Ikeda, M.; Miyake, Y.; Nishibayashi, Y. Angew. Chem., Int. Ed. **2010**, 49, 7289. (b) Yoshida, A.; Ikeda, M.; Hattori, G.; Miyake, Y.; Nishibayashi, Y. Org. Lett. **2011**, 13, 592. (c) Yoshida, A.; Hattori, G.; Miyake, Y.; Nishibayashi, Y. Org. Lett. **2011**, 13, 2460. (d) Ikeda, M.;

Miyake, Y.; Nishibayashi, Y. Chem.—Eur. J. 2012, 18, 3321. (e) Ikeda, M.; Miyake, Y.; Nishibayashi, Y. Organometallics 2012, 31, 3810. (f) Motoyama, K.; Ikeda, M.; Miyake, Y.; Nishibayashi, Y. Organometallics 2012, 31, 3426.

- (11) (a) Zhang, C.; Hu, X.-H.; Wang, Y.-H.; Zheng, Z.; Xu, J.; Hu, X.-P. J. Am. Chem. Soc. 2012, 134, 9585. (b) Han, F.-Z.; Zhu, F.-L.; Wang, Y.-H.; Zou, Y.; Hu, X.-H.; Chen, S.; Hu, X.-P. Org. Lett. 2014, 16, 588. (c) Zhang, D.-Y.; Zhu, F.-L.; Wang, Y.-H.; Hu, X.-H.; Chen, S.; Hou, C.-J.; Hu, X.-P. Chem. Commun. 2014, 50, 14459. (d) Zhou, Y.; Zhu, F.-L.; Duan, Z.-C.; Wang, Y.-H.; Zhang, D.-Y.; Cao, Z.; Zheng, Z.; Hu, X.-P. Tetrahedron Lett. 2014, 55, 2033. (e) Zhang, D.-Y.; Hu, X.-P. Tetrahedron Lett. 2014, DOI: 10.1016/j.tet-let.2014.11.112.
- (12) (a) Detz, R. J.; Delville, M. M. E.; Hiemstra, H.; van Maarseveen, J. H. *Angew. Chem., Int. Ed.* **2008**, 47, 3777. (b) Detz, R. J.; Abiri, Z.; le Griel, R.; Hiemstra, H.; van Maarseveen, J. H. *Chem.—Eur. J.* **2011**, 17, 5921.
- (13) (a) El-Sepelgy, O.; Haseloff, S.; Alamsetti, S. K.; Schneider, C. Angew. Chem., Int. Ed. 2014, 53, 7923. (b) Saha, S.; Schneider, C. Chem. Eur. J. 2015, DOI: 10.1002/chem.201406044. (c) Saha, S.; Alamsetti, S. K.; Schneider, C. Chem. Commun. 2015, 51, 1461–1464. (14) Reviews: (a) Akiyama, T. Chem. Rev. 2007, 107, 5744. (b) Doyle, A. G.; Jacobsen, E. N. Chem. Rev. 2007, 107, 5713. (c) Terada, M. Chem. Commun. 2008, 4097. (d) Zamfir, A.; Schenker, S.; Freund, M.; Tsogoeva, S. B. Org. Biomol. Chem. 2010, 8, 5262. (e) Kampen, D.; Reisinger, C. M.; List, B. Top. Curr. Chem. 2010, 291, 395. (f) Parmar, D.; Sugiono, E.; Raja, S.; Rueping, M. Chem. Rev. 2014, 114, 9047.
- (15) Reviews: (a) van de Water, R. W.; Pettus, T. R. R. Tetrahedron 2002, 58, 5367. (b) Pathak, T. P.; Sigman, M. S. J. Org. Chem. 2011, 76, 9210. (c) Willis, N. J.; Bray, C. D. Chem.—Eur. J. 2012, 18, 9160. (d) Bai, W.-J.; David, J. G.; Feng, Z.-G; Weaver, M. G.; Wu, K.-L.; Pettus, T. R. R. Acc. Chem. Res. 2014, 47, 3655—3664. (e) Singh, M. S.; Nagaraju, A.; Anand, N.; Chowdhury, S. RSC Adv. 2014, 4, 55924.
- (16) (a) Zhang, Y.; Sigman, M. S. J. Am. Chem. Soc. 2007, 129, 3076. (b) Alden-Danforth; Scerba, M. T.; Lectka, T. Org. Lett. 2008, 10, 4951. (c) Jensen, K. H.; Pathak, T. P.; Zhang, Y.; Sigman, M. S. J. Am. Chem. Soc. 2009, 131, 17074. (d) Lv, H.; You, L.; Ye, S. Adv. Synth. Catal. 2009, 351, 2822. (e) Pathak, T. P.; Gligorich, K. M.; Welm, B. E.; Sigman, M. S. J. Am. Chem. Soc. 2010, 132, 7870. (f) Wilcke, D.; Herdtweck, E.; Bach, T. Synlett 2011, 1235. (g) Rueping, M.; Uria, U.; Lin, M. Y.; Atodiresei, I. J. Am. Chem. Soc. 2011, 133, 3732. (h) Jana, R.; Pathak, T. P.; Jensen, K. H.; Sigman, M. S. Org. Lett. 2012, 14, 4074. (i) Luan, Y.; Schaus, S. E. J. Am. Chem. Soc. 2012, 134, 19965. (j) Lv, H.; Jia, W. Q.; Sun, L. H.; Ye, S. Angew. Chem., Int. Ed. 2013, 52, 8607. (k) Izquierdo, J.; Orue, A.; Scheidt, K. A. J. Am. Chem. Soc. 2013, 135, 10634. (l) Lee, A.; Younai, A.; Price, C. K.; Izquierdo, J.; Mishra, R. K.; Scheidt, K. A. J. Am. Chem. Soc. 2014, 136, 10589. (m) Zhao, W.; Wang, Z.; Chu, B.; Sun, J. Angew. Chem., Int. Ed. 2014, DOI: 10.1002/anie.201405252. (n) Hsiao, C. C.; Liao, H. H.; Rueping, M. Angew. Chem., Int. Ed. 2014, 53, 13258. (o) Wang, Z.; Ai, F.; Wang, Z.; Zhao, W.; Zhu, G.; Lin, Z.; Sun, J. J. Am. Chem. Soc. 2015, 137, 383.
- (17) CCDC 1038657 and CCDC 1036085 contain the supplementary crystallographic data of **4e** and 7 for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre (CCDC) via www.ccdc.cam.ac.uk/data_request/cif. For details concerning the crystal structure of **4e** and **7**, see the Supporting Information as well.
- (18) Reaction of the pinacolone-derived enamide with propargylic alcohol $\mathbf{1a}$ (\mathbf{R}^1 , \mathbf{R}^2 = H) furnished the product in 74% yield, 96:4 dr, and 57:43 er.